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### Two-Photon Fluorescence Property of Tris(4,7-Diphenyl-1,10-phenanthroline)ruthenium(II)perchlorate

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## Two-Photon Fluorescence Property of Tris(4,7-Diphenyl-1,10- phenanthroline)ruthenium(II)perchlorate

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The two-photon absorption (TPA) characteristics of an octupolar metal complex; i.e., tris(4,7-diphenyl-1,10-phenanthroline)ruthenium(II)per-chlorate (Ru(dpp)<sub>3</sub>), have been investigated. The TPA cross-section was evaluated by means of a fluorescence-based technique. As a result, the TPA cross-section of Ru(dpp)<sub>3</sub> was estimated to be 210 cm<sup>4</sup>/photon at the wavelength of 830 nm.

**Keywords:** two-photon absorption; two-photon emission; octupolar molecule; metal complex; tris(4,7-diphenyl-1,10-phenanthroline)-ruthenium(II)perchlorate

### INTRODUCTION

A great deal of effort is being placed in developing materials with large two-photon absorption (TPA) cross sections. A material with a large TPA cross-section is potentially utilized for various photonics applications such as 3D-fluorescence imaging, 3D optical data storage, 3D lithographic microfabrication, photodynamic cancer therapy, optical limiting, and so forth<sup>[1]</sup>. A metal complex has been considered to

exhibit a large nonlinear optical (NLO) effect owing to its metal-to-ligand charge transfer (MLCT) or ligand-to-metal charge transfer (LMCT) interaction<sup>[2]</sup>. Since the TPA phenomena is one of the NLO effects, the TPA cross-section of a metal complex is expected to be considerably large. It is recently reported that the TPA cross-section of an octupolar molecule is large<sup>[3]</sup>. Therefore, we have undertaken to study the TPA characteristics of an octupolar metal complex fluorophore, i.e., tris(4,7-diphenyl-1,10-phenanthroline)ruthenium(II)-perchlorate ( $\text{Ru}(\text{dpp})_3$ ; Figure 1). Then the TPA cross-section of  $\text{Ru}(\text{dpp})_3$  has been evaluated by means of a fluorescence-based technique. As a result, the TPA cross-section of  $\text{Ru}(\text{dpp})_3$  was estimated to be  $210 \text{ cm}^4/\text{s/photon}$  at the wavelength of 830 nm.

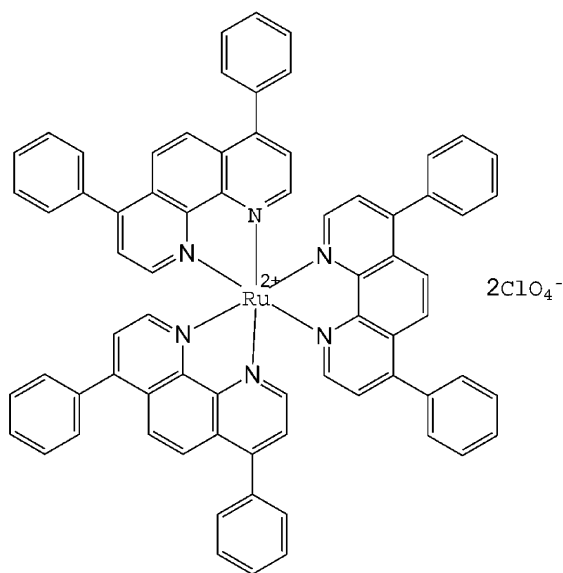


FIGURE 1 Chemical Formula of  $[\text{Ru}(\text{dpp})_3](\text{ClO}_4)_2$ .

## EXPERIMENT

$\text{Ru(dpp)}_3$  was prepared by the same way reported in the literature<sup>[5]</sup>. The TPA cross-section of  $\text{Ru(dpp)}_3$  was determined in a way similar to the one reported in the literature<sup>[4]</sup>.  $\text{Ru(dpp)}_3$  was solved in chloroform at the concentration of  $10^{-2}\text{M}$ . A relative magnitude of TPA cross-section of  $\text{Ru(dpp)}_3$  was determined by comparing the observed TPA spectrum with that of a reference sample of ethanol solution of rhodamine-B. An optical parametric amplifier (LAMBDA, Scan-Mate OPA) with a pulse duration of 2 nsec, a repetition rate of 10 Hz and an energy of 0.5 mJ was employed for excitation.

## RESULTS AND DISCUSSION

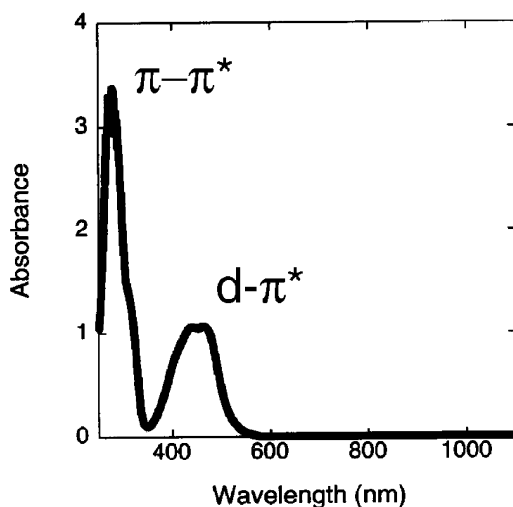


FIGURE 2 The absorption spectrum ( $10^{-6}\text{ M}$  chloroform solution) of  $\text{Ru(dpp)}_3$ .

The absorption spectrum (Figure 2) shows that Ru(dpp)<sub>3</sub> is transparent at the excitation wavelength range. Nevertheless, as shown in Figure 3, emission was observed. As can be seen from Figure 2, weak d- $\pi^*$  transition absorption band is observed at the half of the wavelength we have excited. This fact indicates that the observed emission is the two-photon emission related to the d- $\pi^*$  transition.

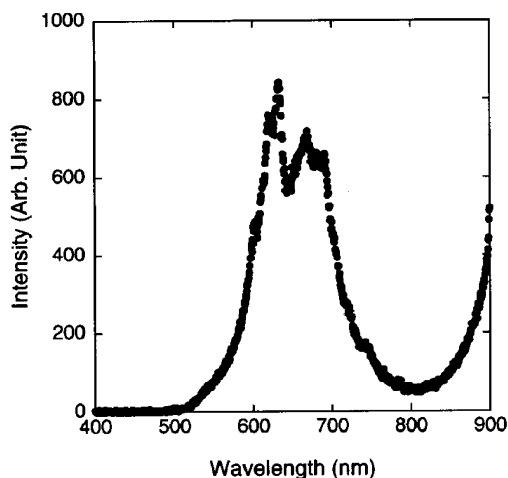


FIGURE 3 The two-photon emission spectrum (excited at the wavelength of 900 nm,  $10^{-2}$  M chloroform solution) of Ru(dpp)<sub>3</sub>.

We measured the dependence of the two-photon emission spectra on excitation wavelength. The result was shown in Figure 4. The TPA cross-section peak is blue shifted with respect to the one-photon absorption peak at twice the wavelength. This behavior is typical for a noncentrosymmetric molecule. The cause of the blue shift is considered that the parity selection rules favor excitation to higher energy levels than do the respective one-photon induced transitions<sup>[6]</sup>.

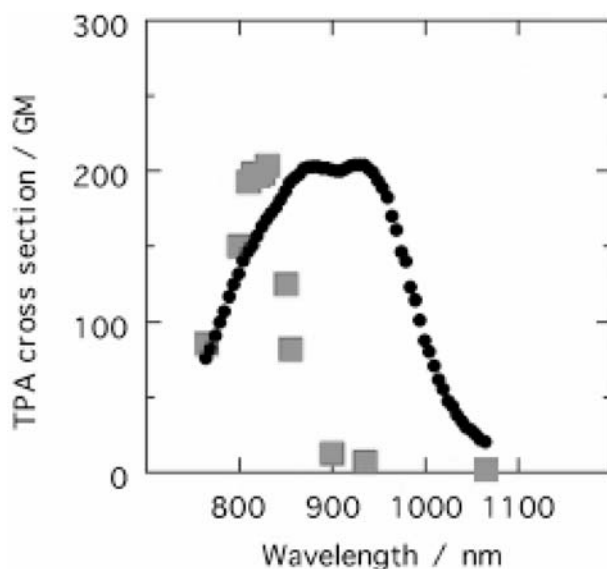


FIGURE 4. Plot of the TPA cross-section (■) compared with the one photon absorption spectra (●) of  $\text{Ru(dpp)}_3$  dissolved in chloroform. Vertical-axis units: arbitrary for one photon absorption spectra and GM ( $1 \text{ GM} = 1 \times 10^{-50} \text{ cm}^4/\text{s/photon}$ ) for the TPA spectra. Horizontal-axis units: wavelengths as shown for the TPA spectra; the one-photon absorption spectra wavelengths have been doubled as plotted.

The maximum value of the TPA cross-section of  $\text{Ru(dpp)}_3$  was estimated to be  $210 \times 10^{-50} \text{ cm}^4/\text{s/photon}$  at the excitation wavelength of 830 nm. As shown in Figure 2, the absorbance concerning the  $d-\pi^*$  transition is quite smaller than that of the  $\pi-\pi^*$  transition. This fact indicates that the degree of MLCT of  $\text{Ru(dpp)}_3$  is not so large. Nevertheless, the TPA cross-section of  $\text{Ru(dpp)}_3$  observed at the twice wavelength of the  $d-\pi^*$  transition is larger than that of the well-known TPA compounds<sup>[4]</sup>. It is often pointed out that a molecule with large

CT degree exhibits large NLO effects<sup>[2]</sup>. Thus, it is natural to consider that an octupolar metal complex which has a large MLCT or LMCT exhibits very large TPA cross-section. Consequently, it should be concluded that an octupolar metal complex like Ru(dpp)<sub>3</sub> exhibits promising TPA characteristics.

## CONCLUDING REMARKS

We evaluated the two-photon absorption (TPA) cross-section of the octupolar metal complex fluorophore (Ru-(dpp)<sub>3</sub>) by means of a fluorescence-based technique. As a result, we have found that an octupolar metal complex like Ru(dpp)<sub>3</sub> exhibits promising TPA characteristics. In order to obtain a metal complex with a larger TPA cross-section than that of Ru(dpp)<sub>3</sub>, enhancement of the oscillator strength should be effective. The further research along with the above principle is now in progress.

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